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NEW ANTIBIOTICS, CARBAZOMYCINS A AND B

III. TAXONOMY AND BIOSYNTHESIS

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The carbazomycin-producing microorganism, strain H 1051-MY 10, was determined to a strain of *Streptoverticillium ehimense*.

Biosynthesis of carbazomycin B was studied using ¹⁴C-labeled and ¹³C-enriched precursors in combination with ¹³C NMR spectroscopy. The C-2 carbon of [2-¹³C]tryptophan was shown to be involved at the C-3 carbon in carbazomycin B and both carbons of [1,2-¹³C]acetate at the C-1 and C-10 moiety of the antibiotic.

 $[CH_{3}^{-13}C]$ Methionine was involved at the methoxyl group but not at the methyl group on the C-2 carbon of the antibiotic. Neither of the labeled carbons, $[1^{-14}C]$ tryptophan nor $[2,3^{-13}C]$ propionic acid, was detected in the antibiotic, and a progenitor of the C-2 and C-11 moiety of the antibiotic has not been determined.

Carbazomycins A and B were produced by an unidentified microorganism, tentatively designated as strain H 1051-MY 10. Carbazomycins inhibited mainly the growth of phytopathogenic fungi and also showed weak antibacterial and antiyeast activities¹). The structure of carbazomycin B (I) was determined to be 4-hydroxy-3-methoxy-1,2-dimethylcarbazole by ¹H and ¹³C NMR spectra²) and by X-ray crystallographic analysis⁸). Consequently, the structure of carbazomycin A (II) was postulated to be 3,4-dimethoxy-1,2-dimethylcarbazole^{1,2}). Thus, carbazomycins are the first antibiotics shown to contain carbazole nucleus¹).

Taxonomy of the carbazomycin-producing microorganism and the biosynthesis of carbazomycin B are reported in this paper.

Taxonomy of Carbazomycin-Producing Microorganism

Strain H 1051-MY 10 was isolated from a soil sample collected in Ni-imi City, Okayama Prefecture, Japan. Strain H 1051-MY 10 grows well on the media recommended in the International Streptomyces Project (ISP)⁴). It produces good aerial mycelium on ISP-media 2, 3 and 4. The color of the aerial mycelium is pale brown with grayish pink tinge. Aerial mycelium is cottony, and, when examined microscopically, consists of long and branched hyphae bearing many whorls (Biverticulate forms) as shown in Fig. 1. The spore surface is smooth as shown in Fig. 2. Morphological characteristics of strain H 1051-MY 10, place it in the genus *Streptoverticillium*. For microscopic observation, an agar medium consisting of 1% soluble starch or maltose, 0.2% yeast extract and 1.5% agar (pH 7.0) is superior to ISP-media.

	Growth	Aerial mycelium	Reverse side of colony	Soluble pigment
Yeast extract - malt extract agar (ISP medium 2)	Good	Good Pale brownish yellow [3 ba]	Brown [3 pi~4 pi]	Light brown
Oatmeal agar (ISP medium 3)	Good	Good Pale brown with pink tint [3 cb]	Brown [3 le]	Light brown
Inorganic salts - starch agar (ISP medium 4)	Good	Good Pale brown with pink tint [3 cb]	Brown [3 ie~3 lg]	Pale brown
Glycerol - asparagine agar (ISP medium 5)	Moderate	Poor White to pale brownish yellow [3 ba]	Pale pinkish yellow [3 ca]	None

Table 1. Cultural characteristics of strain H 1051-MY 10.

Hue numbers in parentheses were described according to the Color Harmony Manual.⁶⁾

The cultural and physiological characteristics of strain H 1051-MY 10 are shown in Table 1 and Table 2, respectively. The following species were selected from BERGEY's manual⁵⁰ on the basis of cultural and physiological properties, and were compared with strain H 1051-MY 10 according to ISP

Fig. 1. Photomicrograph of strain H 1051-MY 10. (on maltose - yeast extract agar, $\times 100$)



Fig. 2. Electron micrograph of spore chains of strain H 1051-MY 10.



procedures: Streptoverticillium cinnamoneum (ISP 5005), Streptoverticillium hachijoense (ISP 5114), Streptoverticillium kentuckense (ISP 5259) and Streptoverticillium ehimense (ISP 5253). The results indicate that strain H 1051-MY 10 is similar to S. ehimense. Differences observed

Table 2. Physiological characteristics of strain H 1051-MY 10.

Melanin production on Tyrosine agar (ISP medium 7)	Negative	
Peptone - yeast extract iron agar (ISP medium 6)	Positive	
Tryptone yeast extract broth (ISP medium 1)	Weakly positive	
H_2S production	Positive	
Starch hydrolysis	Positive	
Gelatin liquefaction	Positive	
Skim milk		
Coagulation	Negative	
Peptonization	Positive	
Carbon utilization		
D-Glucose	+	
L-Arabinose	-	
D-Xylose	-	
D-Fructose	+	
D-Mannitol	-	
Sucrose		
L-Rhamnose	-	
Raffinose	-	
Cellulose	-	
<i>i</i> -Inositol	+	
Salicin	_	

between these two strains were as follows: the color of aerial mycelium of strain H 1051-MY 10 on ISP medium 4 was lighter $(3 \text{ cb})^{6}$ than that of *S. ehimense* $(5 \text{ cb})^{6}$. *S. ehimense* utilized D-mannitol as a carbon source, but strain H 1051-MY 10 did not.

D-Mannitol utilization by S. *ehimense* is positive according to the ISP description⁴⁾; but is described as trace utilization in Bergey's manual. Hydrogen sulfide production by S. *ehimense* is negative according to Bergey's manual, but was positive in our experiments.

Since the above differences between *S. ehimense* and strain H 1051-MY 10 are not enough to conclude that these two strains are distinct species, strain H 1051-MY 10 was assigned as a strain of *Strepto-verticillium ehimense* H 1051-MY 10.

Biosynthesis of Carbazomycin B

Carbazole alkaloids have been isolated from higher plants (mostly Rutaceae) and have been classified into three groups, namely C_{13} -, C_{13} - and C_{23} -skeleton groups^{7~9)}. Carbazomycins are unique because of the C_{14} -skeleton and the 1,2,3,4-tetrasubstituted carbazole nucleus. Thus, biosynthesis of the antibiotics can be expected to be different from that of the alkaloids of higher plants^{7~9)} in which the contribution of mevalonic acid was proved. Determination of the progenitors of the substituted benzene nucleus is especially interesting from a viewpoint of biosynthesis.

The biosynthesis of I was studied in shake cultures of *S. ehimense* H 1051-MY 10, and the fermentation experiments were performed as described previously¹⁾. Based on time-course studies, the labeled precursors were added to the cultures aseptically at 48 hours after inoculation, and the cultivation was continued for additional 48 hours. About 50 mg/liter of I was recovered by methanol extraction of the mycelial cake followed by column chromatography on silica gel eluted with a mixture of benzene acetone (30: 1). For radioactivity measurement, the methanol extract of the mycelial cake from one flask was evaporated to dryness and separated by preparative thin-layer chromatography (TLC) on silica gel developed with a mixture of benzene - acetone (10: 1). The spot of I was quantitated by TLCchromatoscanner (Shimadzu CS-920) at 245 nm. The spot was scraped off, eluted with methanol, evaporated to dryness and dissolved in a toluene-based scintillation fluid to measure radioactivity. For ¹⁸C NMR measurements, I was purified by silica gel column chromatography of the extract of the mycelial cake using a mixture of benzene - acetone (30: 1) as the eluent. Recrystallization from a benzene *n*-hexane mixture yielded yellow prisms with a melting point of 160~162°C.

The structure of I strongly suggests that tryptophan and a C-1 unit would be logical precursors. In our first experiments with ¹⁴C-labeled precursors, DL-[3-¹⁴C]tryptophan, L-[1-¹⁴C]tryptophan, L-[CH₃-¹⁴C]methionine, sodium[1-¹⁴C]acetate and sodium[2-¹⁴C]acetate were added to the culture medium composed of 1.5% soluble starch, 1.0% glucose, 2.0% soy bean meal, 0.5% Ebios (dried yeast, distributed by Tanabe Pharmaceutical Co., Ltd.), 0.25% NaCl and 0.3% CaCO₃ (pH 7.6 before sterilization).

As shown in Table 3, DL-[3-14C]tryptophan and L-[CH₃-13C]methionine gave good total in-

Table	3.	Incorporation	of ¹⁴ C-labelled	precursors	to
carb	azo	mycin B (I).			

Labelled compounds	Radio- activity fed (µCi/ flask)	Yield (µmole/ flask)	Total incor- poration (%)
DL-[3-14C]-			
Tryptophan	2.5	22	14.7
L-[¹⁴ CH ₃]- Methionine	2.5	15	1.9
Sodium[1-14C]- acetate	2.5	17	0.12
Sodium[2- ¹⁴ C]- acetate	2.5	14	0.24
L-[1- ¹⁴ C]- Tryptophan	2.5	20	0.0

corporations, 14.7% and 1.9%, respectively, into I. While, $[1^{-14}C]$ and $[2^{-14}C]$ acetates gave lower incorporations of 0.12% and 0.24%, respectively, and L- $[1^{-14}C]$ tryptophan was not incorporated at all. These high incorporation data prompted us to carry out feeding experiments with ¹³C-enriched compounds to locate the labeled carbons. By feeding L- $[CH_3^{-18}C]$ methionine, isolated I showed the ¹³C NMR spectrum only enriched at the carbon signal due to the *O*-methyl group as seen in Fig. 3. DL- $[2^{-13}C]$ Tryptophan mixed with DL- $[3^{-14}C]$ tryptophan (already known to be involved) was fed to obtain radioactive I in 12.1% incorporation ratio. The ¹⁸C NMR spectrum of the compound in acetone- d_6 showed the significant increment of the signal of C-3 (δ 139.3, singlet) as shown in Fig. 4. This fact proved the contribution of tryptophan to C-3 and C-4 of the hexa-substituted benzene ring besides the indole ring. Though the incorporation ratios of radioactive acetates were not very high, doubly labeled sodium[1,2⁻¹³C]acetate was fed expecting the appearance of satellite signals due to ¹³C-¹³C coupling. As seen in Fig. 5, the ¹³C

*sp*² coupling constant.

As for remaining two carbons, namely C-2 and C-11, suitable precursors are not yet found. Plau-



NMR signals of C-1 and C-10 of 1 appeared along with satellite peaks of J=46 Hz, reasonable for sp^{s} -







Fig. 5. ¹³C NMR spectrum of carbazomycin B derived from sodium[1,2-¹³C]acetate (in CDCl₃).



Fig. 6. Biosynthetic scheme for carbazomycin B.



sible candidates, such as [1,2- or $3^{-14}C$]propionate, [$1^{-14}C$]glycolate, [$2^{-14}C$]pyruvate and L-[$U^{-14}C$]alanine gave negative results.

Although the origin of the C_2 -part (C-2 and C-11) of I has not been clarified, the present study has indicated a biosynthetic scheme for I, where tryptophan reacts with acetate and an unknown unit after decarboxylation and deamination, followed by *O*-methylation with methionine as shown in Fig. 6.

During our study of biosynthesis, it was concluded that the assignments of ¹³C NMR of I and II previously reported^{2, 8)} should be revised as indicated in Table 4. The ¹H NMR (270 MHz) assignments in acetone- d_{θ} for H-5, H-6, H-7 and H-8 in I are δ 8.25, 7.11, 7.27 and 7.41, respectively. Then, C-5, C-6, C-7 and C-8 of I were assigned by proton selective decoupling experiments. The C-1 signal previously assigned to δ 110.0 (overlapped with C-8) should be assigned to δ 109.3 (overlapped with C-4a), because previous assignments were incorrect due to misreading of multiplicity of off-resonance spectra. When the spectrum was measured in acetone- d_{θ} , there appeared two singlet peaks at δ 110.6 (C-4a) and δ 109.7 (C-1), and a doublet at δ 111.0 (C-8). Two C-methyl signals at δ 12.7 and 13.1 had previously been assigned to C-10 and C-11, respectively. In the double labeled experiments, as already described, satellite signals of I were observed at the lower methyl signal (δ 13.1) and at the C-1 (δ 109.3) which was quite distinct from that of C-2 (δ 127.0). Further, the lower methyl signal in the ¹H NMR (δ 2.37, 10- CH_{3}) of I corresponded with the lower carbon signal (δ 13.3) in acetone- d_{6} by proton selective decoupling experiments at 270 MHz (67.5 MHz for ¹³C). Thus, previous assignments of two C-methyl signals should be reversed. The previous assignments of C-3 (δ 142.0) and C-4 (δ 138.5) should also be reversed as shown in Table 4. When $[2^{-13}C]$ tryptophan was fed, the signal at δ 138.5 (δ 139.3 in acetone- d_{θ}) of I was enlarged. This labeled compound of I was methylated with dimethyl sulfate to obtain labeled II. The ¹³C NMR of labeled II showed an enlarged peak at δ 144.4 ($\Delta \delta$ +3.9). This substitution-induced

Table 4. Revised assignments of ¹³C-chemical shifts (δ) of carbazomycins A and B.



Carbazomycin B: R=H(I)Carbazomycin A: $R=CH_{3}(II)$

Carbon No.	Carbazomycin B	Carbazomycin A
C-1	109.3 (109.7)	113.5
C-2	127.0 (127.8)	128.7
C-3	138.5 (139.3)	144.4
C-4	142.0 (143.6)	145.9
C-4a	109.3 (110.6)	114.4
C-4b	123.3 (124.3)	122.8
C-5	122.7 (123.1)	122.5
C-6	119.5 (119.2)	119.4
C-7	124.7 (124.8)	125.0
C-8	110.0 (111.0)	110.3
C-8a	139.3 (140.7)	139.4
C-9a	136.8 (137.9)	136.4
C-10 (1-CH ₃)	13.1 (13.3)	13.6
C-11 (2-CH ₃)	12.7 (12.8)	12.6
C-12 (3-OCH ₃)	61.4 (61.3)	61.1*
C-13 (4-OCH ₃)		60.5*

Spectra were taken with JEOL PFT-100 spectrometer equipped with EC-6 computer at 25.15 MHz using tetramethylsilane as an internal standard and CDCl₃ as a solvent. Values in parentheses were taken in acetone- d_{θ} . Parameters are as follows: spectral width 5 KHz, pulse width 16 μ seconds (45°), repetition time 2 seconds, computer limited resolution ± 0.1 ppm.

* May be reversed.

Fig. 7. Substitution-induced shifts of aromatic carbons by *O*-methylation.



shift value ($\Delta\delta$) was compared with that of a model compound, 1,2,3-trimethoxybenzene (III)¹⁰). As shown in Fig. 7, on going from 1, 3-dimethyl-2hydroxybenzene (IV), C-1 and C-3 carbons of III showed the largest $\Delta\delta$ value (+6.3), while the substituted position (C-2) showed smaller $\Delta\delta$ value (+3.4). Further, the *para*-carbon (C-5) of III showed fairly large $\Delta\delta$ value (+4.9). In the previous assignments of ¹⁸C NMR of I, all signals except C-3 and C-4 were in good accordance with this shift trend. Thus, assignments of C-3 and C-4 can only be explained by being reversed.

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